DIPTAMINE - A NEW ALKALOID FROM THE EPIGEAL PART OF Dipthychocarpus strictus

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Continuing the preparation of the total alkaloids of the epigeal part of Dipthychocarpus strictus (Fisch.) Trautv. (family Cruciferae) collected in the Chimkent province, when the mixture of bases was chromatographed on a column of silica gel (1:25), the chloroform fractions yielded a crystalline substance with mp 87-89°C. The base proved to be new and has been named diptamine. Diptamine (I) is an optically active substance, $[\alpha]_D^{20}$ -42° (MeOH), readily soluble in chloroform, methanol, and water and sparingly soluble in benzene, ether, and acetone.

The IR spectrum of the alkaloid shows absorption bands at (cm⁻¹) 3320-3370 (active hydrogen), 1630 (amide carbonyl), and 1035 (sulfoxide group). The mass spectrum of (I) contains the peaks of ions with m/z 262 (M⁺)(5), 247 (M - CH₃)⁺(11), 204(M - NH-CH(CH₃)₂⁺(30), 199 - $_{_{3}}^{O}$

 $_{(M-S-CH_3)^+(25)}$ 161(10), 132(28), 114(16), 101(7), 84(27), 71(60), 69(52), 58(100%). In a comparative study of the mass spectra of diptocarpamine (II) [1] and (I), it can be seen that they differ in the m/z value of the molecular ion by 14 m.u.

The PMR spectrum of diptamine (CDCl₃) showed the signals of the protons of the following groups: $C(CH_3)_2$ (1.13 ppm, 6 H, d, J = 8 Hz; $(CH_2)_5$ (1.25-1.75 ppm, 10 H, m); $CH_3S \rightarrow O(2,52;$ 3 H, s); $OCS \rightarrow CCH_2 \rightarrow CC$

The presence in the mass spectrum of (I) of a strong peak of an ion with m/z 58 as in the mass spectrum of diisopropylurea [2] shows that diptamine is also a N-alkyl derivative of isopropylurea. The difference of 14 m.u. between the molecular weights of (I) and (II) shows that diptamine is a homolog of diptocarpamine differing from the latter by one methylene group.

Thus, on the basis of spectral characteristics and the comparative study with diptocarpamine, diptamine has the structure of N-isopropyl-N'-(7-methylsulfinyl-n-heptyl)urea.

LITERATURE CITED

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